## Lesson Plans $\rightarrow$ Apr 15<sup>th</sup> – Apr 19<sup>th</sup> Mrs. Misty Thompson

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Standards	Objectives	Activities	Evaluation	Accommodations
CHEM.B.2.1.5 Balance chemical equations by applying the Law of Conservation of Matter CHEM.B.2.1.2 Use stoichiometric relationships to calculate the amounts of reactants and	<ul> <li>Use gas laws to solve problems</li> <li>Use stoichiometry to determine the volume of a chemical reaction</li> </ul>	Monday—Review answers to practice and complete quiz review Tuesday— Gas law quiz and complete 13.2 notes	Assignments: • Gas law quiz • 13.2 practice • 13.3 practice • Chapter 13 test	<ul> <li>Up front seating</li> <li>Extra time for class assignments if needed</li> <li>One on one tutoring</li> <li>Lesson modification</li> </ul>
products involved in a chemical reaction. <b>CHEM.B.2.2.1</b> Utilize mathematical relationships to predict changes in the number of particles, the temperature, the pressure, and the volume in a gaseous system (i.e., Boyle's law, Charles's law, Dalton's law of partial pressures, the combined gas law, and the ideal gas law).		<ul> <li>Wednesday—Complete 13.2 practice</li> <li>Thursday— Review answers to practice and 13.3 notes. 13.3 practice</li> <li>Friday—Review answers to practice and work on test review</li> </ul>	<ul> <li>Homework:</li> <li>Online assignments:</li> <li>→ 13.1 online due Monday by 11:00 pm</li> <li>→ 13.2 online review due Thursday by 11:00 pm</li> <li>Finish any classwork assignments</li> <li>Participation in class discussion.</li> </ul>	as needed

## Mrs. Thompson's Daily Class Schedule

Rm. 202	Period 1	ROCK Room	8:05-8:47
Rm. 202	Period 2	8 <sup>th</sup> grade Science	8:50-9:30
Rm. 202	Period 3	8 <sup>th</sup> grade Science	9:33-10:13
Rm. 202	Period 4	Planning	10:16-10:56
		Lunch A	10:56-11:26
Rm. 202	Period 5	8 <sup>th</sup> grade Science	11:29-12:09
Rm. 202	Period 6	Lunch Duty C	12:22-12:52
Rm. 202	Period 7	Chemistry	12:55-1:35
Rm. 202	Period 8	Chemistry Lab	1:38-2:18
Rm. 202/204	Period 9	Homeroom/FFA Club	2:21-2:51